SPECTRAL DATA ON NEW COMPOUNDS

Ultraviolet Wave lengths of maxima and inflections (*) in 95% ethanol are given in m µ							_	
						Infrared Frequencies in Cm. ⁻¹		
_	•	with intensiti	ies in $(\log \epsilon)$		_	\mathbf{Medium}	NH	NO ₂ (Aliphatic)
I	275	296*		307*		CHCl ₃	3520	1558, 1387
	(3.83)	(3.70)		(3.54)		Nujol	3380	1553, 1385
II	275	295*		308*		CHCl_3	3440	1552, 1380
	(3.85)	(3.74)		(3.60)		Nujol	3400	1550, 1380
III	275	295*		309*		CHCl_3	3450	1550, 1387 or 136
	(3.81)	(3.73)		(3.57)		Nujol	3390	1547, 1385 or 135
Ia	275*´	297 ´		3 09	357	Nujol	3400	,
	(3.92)	(3.85)		(3.85)	(4.20)	•		
IIa	277*	$\hat{2}97$	302*	309	357	Nujol	3360	
	(3.91)	(3.87)	(3.87)	(3.87)	(4.20)	•		
IIIa	277*	298	304*	309	357	Nujol	3400	
	(3.90)	(3.88)	(3.87)	(3.89)	(4.20)	U		

232° dec., calcd. for $C_{23}H_{21}N_5O_8$ (495.44): C, 55.75; H, 4.27; N, 14.14; found: C, 55.69; H, 4.58; N, 14.07; IIa (94% yield from II), red crystals from ethanol, m.p. 176-176.5°, calcd. for $C_{29}H_{25}N_5O_8$ (571.53): C, 60.94; H, 4.41; N, 12.25; found: C, 60.96; H, 4.74; N, 12.08; IIIa (62% yield from III), bright red crystals from ethanol, m.p. 213-215° calcd. for C₃₀H₂₇N₅O₈ (585.56): C, 61.53; H, 4.65; N, 11.96; found: C, 61.83; H, 4.91; N, 12.09. The tryptamine from I was also characterized as the hydrochloride, which had m.p. 245-247° dec., in agreement with that reported, 248-250°.6

Hydrogenation of I at 2 atm. over 10% palladium on charcoal catalyst, resulting in concomitant reduction of the nitro group and debenzylation, gave serotonin in 69% yield as the hygroscopic creatinine sulfate monohydrate salt, m.p. 212-214°, mixed m.p. with an authentic sample (of m.p. 214-216°), 212-216°. The infrared spectra of the two samples in Nujol were identical and in agreement with the spectrum described in the literature $-\lambda_{max}$ in water: 220 m μ (log ϵ 4.40), 274 (3.72), 293 inflection (3.63). Reported λ_{max} in water: 275 m μ and 293 inflection. Calcd. for $C_{14}H_{23}N_5O_7S$ (405.43): C, 41.47; H, 5.72; N, 17.28; S, 7.91; found: C, 41.43; H, 5.73; N, 17.54; S, 8.14. This new synthesis of serotonin in two steps from 5-benzyloxyindole is simpler than previously described methods. The over-all yield (31%) from 5-benzyloxyindole appears to be higher than that reported for all other methods except those of Speeter and Anthony8 (probably greater than 60%) and Young (34%).9

SCHOOL OF CHEMISTRY University of Minnesota MINNEAPOLIS 14, MINN.

WAYLAND E. NOLAND ROBERT A. HOVDEN¹⁰

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Formylation of Aromatic Amines with Dimethylformamide¹

Sir:

A variety of methods are available for the preparation of formamides.2 However, only several procedures exist for the direct formylation of amines with more readily available formamides.

In 1886, Just described the formylation of phenylhydrazine with formamide at 130°.3 Several years later Hirst and Cohen obtained formanilides from a number of aromatic amines using formamide in glacial acetic acid. A more recent procedure employs the amine hydrochloride and formamide.5

We now wish to report the facile formulation of aromatic amines with dimethylformamide in the presence of sodium methoxide. Conversion to the formanilide is accomplished by heating a mixture of sodium methoxide (0.3 mole), the aniline (0.15 mole) and dimethylformamide (150 ml.) at reflux for 30 min. The reaction is accompanied by the evolution of dimethylformamine. Generally, the formanilide, obtained by diluting the reaction mixture with water, does not require further purification. Typical examples include 2-iodoformanilide, colorless needles, m.p. 113-113.5°, 68% yield (Anal. Calcd. for C₇H₆INO: C, 34.04; H, 2.45; N, 5.67.

(1) Support of this investigation by a Frederick Gardner Cottrell grant from the Research Corporation is gratefully acknowledged.

(2) Cf., H. Winteler, A. Bieler, and A. Guyer, Helv. Chim. Acta, 37, 2370 (1954); L. Pichat, M. Audinot, and J. Monnet, Bull. soc. chim. France, 85 (1954); P. de Benneville, J. S. Strong, and V. T. Elkind, J. Org. Chem., 21, 772 (1956); R. Gompper, Chem. Ber., 89, 1762 (1956); G. Olan and S. Kuhn, Chem. Ber., 89, 2211 (1956); E. C. Kornfeld et al., J. Am. Chem. Soc., 78, 3112 (1956); R. M. Roberts and P. J. Vogt, J. Am. Chem. Soc., 78, 4778 (1956); F. E. King, J. W. Clark-Lewis, and R. Wade, J. Chem. Soc., 880 (1957); C. W. Huffman, J. Org. Chem., 23, 727 (1958)

(3) F. Just, Ber., 19, 1201 (1886). At 160-170° this reaction yields 1-phenyl-1,2,4-triazole; M. Sekiya and S. Ishikawa, Yakugaku Zasshi, 78, 549 (1958).

(4) H. R. Hirst and J. B. Cohen, J. Chem. Soc., 67, 830 (1895).

(5) A. Galat and G. Elion, J. Am. Chem. Soc., 65, 1566 (1943).

⁽⁶⁾ M. E. Speeter, R. V. Heinzelmann, and D. I. Weisblat, J. Am. Chem. Soc., 73, 5514 (1951).

⁽⁷⁾ M. M. Rapport, A. A. Green, and I. H. Page, J. Biol. Chem., 176, 1243 (1948).

⁽⁸⁾ M. E. Speeter and W. C. Anthony, J. Am. Chem. Soc., 76, 6208 (1954).
(9) E. H. P. Young, J. Chem. Soc., 3493 (1958).

⁽¹⁰⁾ From the M.S. thesis of Robert A. Hovden, University of Minnesota, February 1959. All melting points were determined on a Kofler micro hot stage.

Found: C,33.71; H, 2.27; N, 5.84), 4-bromoformanilide (70.5% yield, m.p. 117–119°; lit.6 m.p. 119°), 2-chloroformanilide (88.4% yield, m.p. 77.5–78°, lit.7 m.p. 77°), 3-chloro-4-methylformanilide, colorless needles, m.p. 97–97.5°, 68.5% yield (Anal. Calcd. for $C_8H_8ClNO: C$, 56.79; H, 4.77; N, 8.28. Found: C, 56.60; H, 4.66; N, 8.11), and 4-

(6) M. Dennstedt, Ber., 13, 228 (1880).

formamidobenzoic acid (45.6% yield, m.p. 270° dec., lit. 8 m.p. 268° dec.).

The composition of a water labile intermediate and the scope of this formylation reaction are presently under investigation.

DEPARTMENT OF CHEMISTRY UNIVERSITY OF MAINE ORONO, ME. GEORGE R. PETTIT EVAN G. THOMAS

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⁽⁷⁾ F. P. Chattaway and J. P. Orton, Ber., 2, 2396 (1900).

⁽⁸⁾ A. Zehra, Ber., 23, 3625 (1890).